

Original

# Determination of Estrogens in Rat Brains Using Gas Chromatography/Mass Spectrometry/Mass Spectrometry

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Received for Review November 30, 2001. Accepted February 4, 2002.

## Abstract

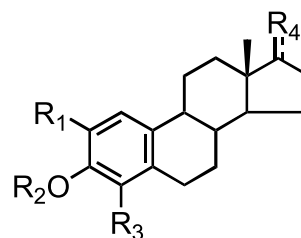
A standard addition determination method of estrone, estradiol and guaiacol estrones in rat brains using gas chromatography/mass spectrometry/mass spectrometry (GC/MS/MS) was developed. Estrone and estradiol were extracted with liquid-liquid extraction, purified successively with silica-gel mini-column chromatography, solid-phase extraction and preparative HPLC. The obtained estrogens were derivatized to dimethylethylsilyl ethers followed by GC/MS/MS analysis. The concentrations of estrone and estradiol were 0-44 and 0-159 pg/g tissue (n=4), respectively. On the other hand, an additional pretreatment procedure was necessary for determination of guaiacol estrones in rat brains. After the liquid-liquid extraction, guaiacol estrones were acetylated with acetic anhydride-pyridine and purified successively with silica-gel mini-column chromatography and solid-phase extraction. The obtained guaiacol estrone acetates were hydrolyzed, derivatized to dimethylethylsilyl ethers and then applied to GC/MS/MS. Although the absolute recovery rates of guaiacol estrones spiked in brain homogenate were higher than 65%, these estrogens in rat brain were not determined.

**Keywords:** estrone, estradiol, guaiacol estrone, rat brain, standard addition method, gas chromatography/mass spectrometry/mass spectrometry

## 1. Introduction

The term "neurosteroids" applies to the steroids that are both synthesized in the nervous system, either *de novo* from cholesterol or from steroid hormone precursors, and that accumulate in the nervous system to levels that are at least in part independent of the steroidogenic gland secretion rates. It has been reported that these steroids are found in larger amounts in brain than in plasma [1]; recently, significant interest has been focused on the biological properties of neurosteroids, such as pregnenolone, dehydroepiandrosterone and their conjugates [2]. Also, the existence of estrogen receptor in brain has been clarified [3], and the usefulness of estrogen-replacement therapy for dementia of Alzheimer type has been reported [4]. Much interest is focused on the action of estrogens on the central nervous system and on the existence of estrogens in the brain. In previous papers, we reported the existence of the classical estrogens {estrone ( $E_1$ ), estradiol ( $E_2$ ) and estriol}, the guaiacol estrones {2-hydroxyestrone 3-methyl ether (2OHE<sub>1</sub>3Me) and 4-hydroxyestrone 3-methyl ether (4OHE<sub>1</sub>3Me)} [5] (Figure 1) and the

catechol estrogens (2- or 4-hydroxy metabolites) [6,7] in rat brains using gas chromatography/mass spectrometry/mass spectrometry (GC/MS/MS) and/or liquid chromatography/mass spectrometry (LC/MS). However, two or three whole brains were used



$E_1$ :  $R_1=R_2=R_3=H$ ,  $R_4=O$   
 $E_2$ :  $R_1=R_2=R_3=H$ ,  $R_4=\beta-OH$   
 2OHE<sub>1</sub>2Me:  $R_1=OMe$ ,  $R_2=R_3=H$ ,  $R_4=O$   
 2OHE<sub>1</sub>3Me:  $R_1=OH$ ,  $R_2=Me$ ,  $R_3=H$ ,  $R_4=O$   
 4OHE<sub>1</sub>3Me:  $R_1=H$ ,  $R_2=Me$ ,  $R_3=OH$ ,  $R_4=O$   
 4OHE<sub>1</sub>4Me:  $R_1=R_2=H$ ,  $R_3=OMe$ ,  $R_4=O$

**Figure 1.** Structures of  $E_1$ ,  $E_2$  and guaiacol estrones.

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for identification at a time; therefore, the quantitative analytical method using part of an entire brain is necessary to clarify the physiological role of these estrogens in the central nervous system.

In this paper, we describe the development of the standard addition determination method of  $E_1$ ,  $E_2$  and guaiacol estrones in part of a rat brain using GC/MS/MS.

## 2. Experimental

### 2.1. Apparatus, materials and animals

GC/MS/MS was performed on a GCQ gas chromatograph/ion trap mass spectrometer (ThermoQuest, San Jose, CA, USA) equipped with a 5% diphenyl-95% dimethylpolysiloxane capillary column (Rtx-5MS, 30m x 0.25mm i.d., 0.25 $\mu$ m df; Restek, Bellefonte, PA, USA), which was used with the following conditions: electron ionization was set at 70 eV and detected with positive ion mode. Carrier gas was helium at a flow rate of 40cm/s. The column oven temperature was set at 60 for 4 min, ramped to 280 at 40 /min and held at 280 for 10min. The split-less injector temperature was set at 60 for 1 min, ramped to 280 at 180 /min and held at 280 for 18 min. The transfer line temperature and the ion source temperature were set at 275 and 200, respectively.

Preparative HPLC was performed on a LC-6A (Shimadzu, Kyoto, Japan) equipped with a J'sphere ODS-H80 (4 $\mu$ m, 150 x 4.6mm i.d.) (YMC, Kyoto) column and an SPD-10A UV detector (280nm) (Shimadzu). The flow rate and the column temperature were set at 1ml/min and 40, respectively.  $E_1$  and  $E_2$  were not separated when MeOH-H<sub>2</sub>O (13:7) was used as the mobile phase, so that both compounds were obtained in the same fraction.

$E_1$  and  $E_2$  were supplied by Teikoku Hormone Mfg. (Tokyo, Japan). Guaiacol estrones (2OHE<sub>1</sub>2Me, 2OHE<sub>1</sub>3Me, 4OHE<sub>1</sub>3Me, 4OHE<sub>1</sub>4Me) and 16-oxoestradiol (16oxoE<sub>2</sub>) {internal standard (IS) for guaiacol estrones} were purchased from Steraloids (Newport, RI, USA). [2,4,6,6,9-<sup>2</sup>H<sub>5</sub>]- $E_1$  and - $E_2$  (ISs for  $E_1$  and  $E_2$ ) [8] were prepared in our laboratories from [6,6,9-<sup>2</sup>H<sub>3</sub>]- $E_1$  and - $E_2$ , respectively, which were kindly supplied by Nippon Kayaku (Tokyo). Silica-gel mini-column chromatography (0.5g, 3 x 0.6cm i. d.) was performed with Silica Gel 60 (70-230mesh; E. Merck, Darmstadt, Germany). ISOLUTE C18 (EC) cartridges (500mg; International Sorbent Technology, Hengoed, Mid Glamorgan, UK) were purchased from Uniflex (Tokyo). Dimethylethylsilylimidazole (DMESI) was obtained from Tokyo Kasei Kogyo (Tokyo), and all the other reagents were analytical grade and commercially available.

Male Wistar strain rats (7 weeks old) were obtained from Japan S.L.C. (Hamamatsu, Japan).

### 2.2. Determination of $E_1$ and $E_2$

Wistar strain rat was decapitated and the entire brain (ca. 1.8 g) was homogenized in 10% EtOH (10 volumes, w/v) using a Teflon homogenizer, and then divided into three parts so that the content of brain in each tube is approximately 0.6 g of tissue. Standard solutions of  $E_1$  and  $E_2$  (each 0, 50, 100 pg in EtOH) and ISs ([<sup>2</sup>H<sub>5</sub>]- $E_1$  and - $E_2$ ; each 1 ng in EtOH) were added to the homogenate. After standing for 10min at room temperature, AcOEt (6ml) was added to the samples and the mixture was then sonicated for 20min in an ultrasonic bath, stirred at room temperature for 30min and centrifuged at 1500g for 20min. The organic layer was collected and the residue was re-extracted with AcOEt (6ml), and the entire organic layer was evaporated under a N<sub>2</sub> gas stream. The residue was dissolved in hexane-AcOEt (20:1, 0.5ml) and subjected to silica-gel mini-column chromatography. After washing with hexane (5ml) and hexane-AcOEt (10:1, 5ml), the fraction eluted with hexane-AcOEt (2:1, 8ml) was evaporated under a N<sub>2</sub> gas stream. The residue was dissolved in 10% MeOH (1ml) and applied to an ISOLUTE C18 (EC) cartridge. After washing with H<sub>2</sub>O (5ml) and 20% MeOH (2 ml), the eluate with MeOH (3ml) was evaporated under a N<sub>2</sub> gas stream. The obtained residue was purified by preparative HPLC and the fraction containing  $E_1$ ,  $E_2$  and ISs (5.5-7.5 min) was evaporated *in vacuo* and the residue was derivatized with DMESI (50 $\mu$ l) for 30min at 60. The reaction mixture was diluted with CHCl<sub>3</sub> (0.5ml), applied to a silica-gel mini-column chromatography, and the eluate with CHCl<sub>3</sub> (5ml) was evaporated

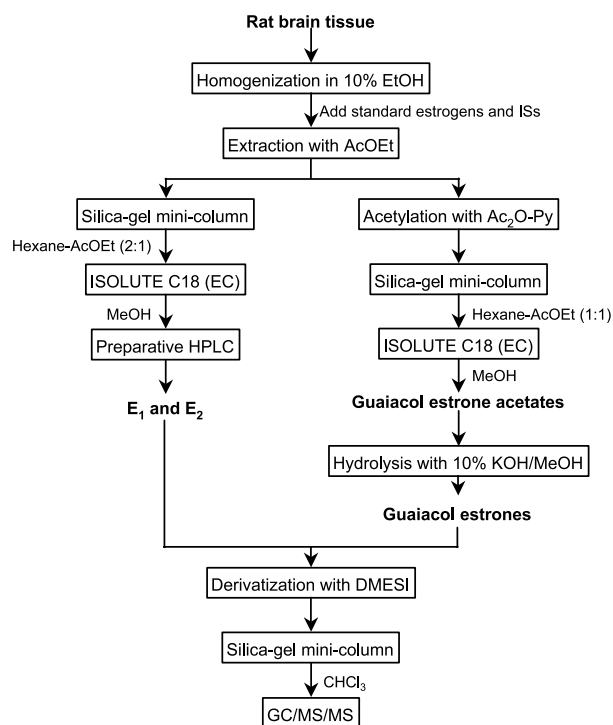


Figure 2. Procedure for determination of  $E_1$ ,  $E_2$  and guaiacol estrones.

under a N<sub>2</sub> gas stream. The residue was dissolved with hexane (50 µl) and an aliquot of which was subjected to GC/MS/MS {precursor ions,  $m/z$  356 and 361 (10–14 min, for E<sub>1</sub> and [<sup>2</sup>H<sub>5</sub>]-E<sub>1</sub>), 444 and 449 (14–18 min, for E<sub>2</sub> and [<sup>2</sup>H<sub>5</sub>]-E<sub>2</sub>); excitation voltage, 1.0 V for E<sub>1</sub> and [<sup>2</sup>H<sub>5</sub>]-E<sub>1</sub> and 1.2 V for E<sub>2</sub> and [<sup>2</sup>H<sub>5</sub>]-E<sub>2</sub>; monitoring ions,  $m/z$  271 (E<sub>1</sub>), 276 ([<sup>2</sup>H<sub>5</sub>]-E<sub>1</sub>), 340 (E<sub>2</sub>) and 344 ([<sup>2</sup>H<sub>5</sub>]-E<sub>2</sub>)} (Figure 2).

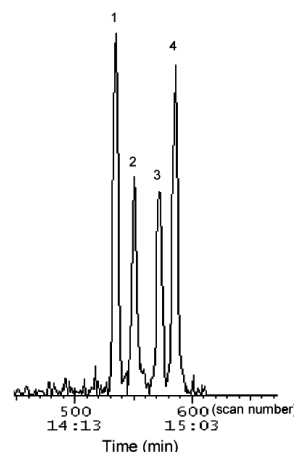
### 2.3. Determination of guaiacol estrones

Standard solutions of guaiacol estrones (each 25–400 pg in EtOH) and IS (16oxoE<sub>2</sub>; 10 ng in EtOH) were added to the Wistar rat brain homogenate corresponding to 0.5 g of tissue, and the estrogens were extracted with AcOEt as described above. After the solvent was evaporated under a N<sub>2</sub> gas stream, the extract was treated with acetic anhydride–pyridine (1:2, 0.3 ml) at room temperature for 30 min. After addition of H<sub>2</sub>O (1 ml), the entire solution was extracted with AcOEt, and the organic layer was successively washed with 5% HCl, 5% NaHCO<sub>3</sub> and H<sub>2</sub>O. After evaporation of the solvent, the residue was dissolved in hexane–AcOEt (10:1, 1 ml), subjected to silica–gel mini–column chromatography and washed with hexane–AcOEt (10:1, 5 ml). The fraction eluted with hexane–AcOEt (1:1, 6 ml) was evaporated under a N<sub>2</sub> gas stream, the residue was dissolved in 10% MeOH (2 ml) and applied to an ISOLUTE C18 (EC) cartridge. After washing with H<sub>2</sub>O (5 ml) and 30% MeOH (2 ml), the eluate with MeOH (3 ml) was evaporated under a N<sub>2</sub> gas stream. The residue was hydrolyzed with 1% KOH/MeOH (0.5 ml) for 30 min at room temperature. After the neutralization with 5% HCl, the entire solution was extracted with AcOEt and the organic layer was washed with H<sub>2</sub>O. The solvent was evaporated under a N<sub>2</sub> gas stream, the obtained residue was derivatized with DMESI, treated as above, and then subjected to GC/MS/MS analysis {precursor ions,  $m/z$  342 (14–16 min, for guaiacol estrones), 429 (16–18 min, for 16oxoE<sub>2</sub>); excitation voltage, 1.0 V; monitoring ions,  $m/z$  257 (guaiacol estrones), 337 (16oxoE<sub>2</sub>)} (Figure 2).

## 3. Results

### 3.1. GC/MS analysis of estrogens

In the previous paper, we chose the methyloxime (MO) and/or trimethylsilyl ether (TMS) as the derivatization form for identification of classical estrogens and guaiacol estrones in rat brain using GC/MS/MS [5]. Although derivatization to MO–TMS was simple and convenient, the complete separation of the four guaiacol estrone isomers was not achieved chromatographically. To find suitable derivatization for E<sub>1</sub>, E<sub>2</sub> and guaiacol estrones, we examined *O*-methylhydroxylamine, *O*-pentafluorobenzylhydroxylamine (for derivatization of the carbonyl group), *N,O*-bis(trimethylsilyl)trifluoroacetamide, DMESI, dimethyl-*iso*-propylsilylimidazole,



**Figure 3.** Product ion mass chromatogram of DMES derivatives of guaiacol estrones (100 pg/tube).

Peaks: 1, 4OHE<sub>1</sub>4Me; 2, 2OHE<sub>1</sub>3Me; 3, 2OHE<sub>1</sub>2Me; 4, 4OHE<sub>1</sub>3Me. GC/MS/MS conditions were described in the experimental section.

heptafluorobutyric anhydride and pentafluorobenzyl bromide (for derivatization of the hydroxy group). Among these derivatization reagents, DMESI appeared to be most promising with respect to resolution, sensitivity and feasibility, and the product ion mass chromatogram was shown in Figure 3.

The mass spectra of DMES derivatives of E<sub>1</sub>, E<sub>2</sub>, 2OHE<sub>1</sub>2Me and 16oxoE<sub>2</sub> showed an ion at  $m/z$  356 ([M]<sup>+</sup>),  $m/z$  444 ([M]<sup>+</sup>),  $m/z$  342 ([M–CH<sub>3</sub>–C<sub>2</sub>H<sub>5</sub>]<sup>+</sup>) and  $m/z$  429 ([M–C<sub>2</sub>H<sub>5</sub>]<sup>+</sup>), respectively, as the base peak ion, so that these ions were chosen as the precursor ions in the GC/MS/MS (Figure 4a). The product ion mass spectra of E<sub>1</sub>– and 2OHE<sub>1</sub>2Me–DMES showed the prominent ion ([precursor ion–85]<sup>+</sup>) at  $m/z$  271 and  $m/z$  257, respectively, which were assigned as the ions formed by dissociation of the steroid C– and D–ring and rearrangement of hydrogen (Figure 4b) [9]. The mass spectra and product ion mass spectra of the other guaiacol estrone isomers did not show such a clear difference from those of 2OHE<sub>1</sub>2Me. The product ion mass spectrum of E<sub>2</sub>–DMES showed ions at  $m/z$  299 and  $m/z$  340. The former product ion was formed by dissociation of the D–ring and the latter ion corresponded to [M–DMESOH]<sup>+</sup>. Although the intensity of the latter ion was lower than that of the former, the ion at  $m/z$  340 was chosen as the monitoring ion because of responsible peak purity. 16oxoE<sub>2</sub>–DMES produced an ion at  $m/z$  337 in GC/MS/MS and it was used as a monitoring ion, but the origin was not assigned. Using these product ions as the monitoring ions, the signal to noise ratio (S/N) of every authentic estrogen at 5 pg/injection was higher than 5.

### 3.2. Determination of E<sub>1</sub> and E<sub>2</sub> in rat brains

A pilot study was performed to optimize the extraction and purification methods. The recovery rates of authentic E<sub>1</sub>, E<sub>2</sub> (50–

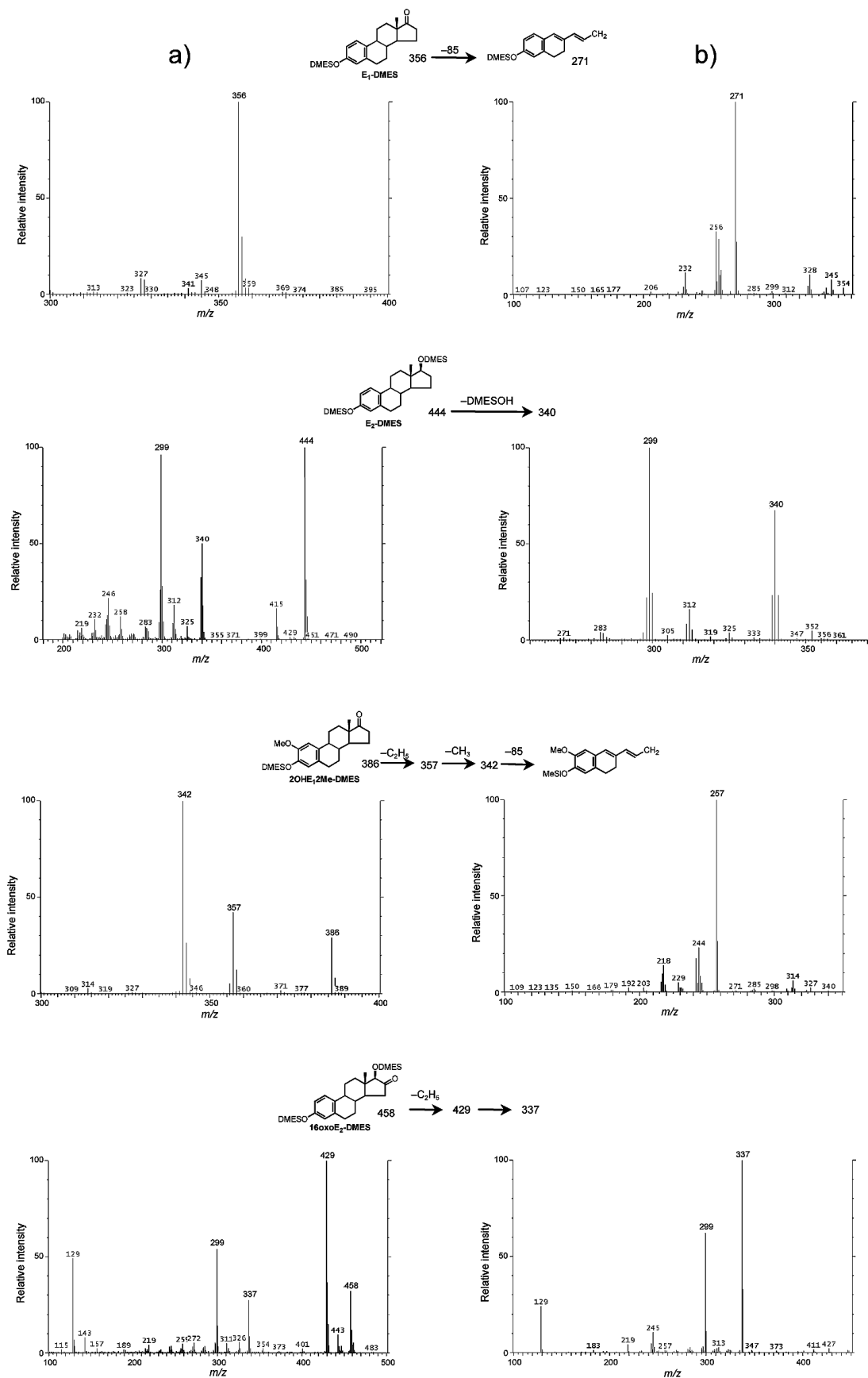


Figure 4. Mass spectra (a) and product ion mass spectra (b) of DMES derivatives of E<sub>1</sub>, E<sub>2</sub>, 2OHE<sub>2</sub>2Me and 16oxoE<sub>2</sub>.

400pg) and their ISs ( $[^2\text{H}_5]$ -E<sub>1</sub> and -E<sub>2</sub>; each 1 ng) from silica-gel mini-column chromatography and solid-phase extraction were around 60–80%. To examine the effect of endogenous substances, ISs were added to the brain homogenate, and then extracted and purified successively by silica-gel mini-column chromatography, solid-phase extraction, and then subjected to GC/MS/MS. The peaks corresponding to the DMES ethers of ISs were detected clearly without interference by other peaks. However, the areas of these peaks tended to decrease as the analyses were repeated owing to endogenous interfering substances; therefore, further purification using preparative HPLC is necessary prior to derivatization to a DMES ether. E<sub>1</sub> and E<sub>2</sub> were separated using the ODS column and MeCN-H<sub>2</sub>O system as a mobile phase. On the other hand, these were not separated using a MeOH-H<sub>2</sub>O system, so that both estrogens and ISs were obtained in the same fraction.

Then the reproducibility and accuracy were examined in order to validate the method. The overall absolute recovery rates were examined using ISs (1 ng) which were added to the brain homogenate. Although the relative standard deviation (RSD) in the same as-

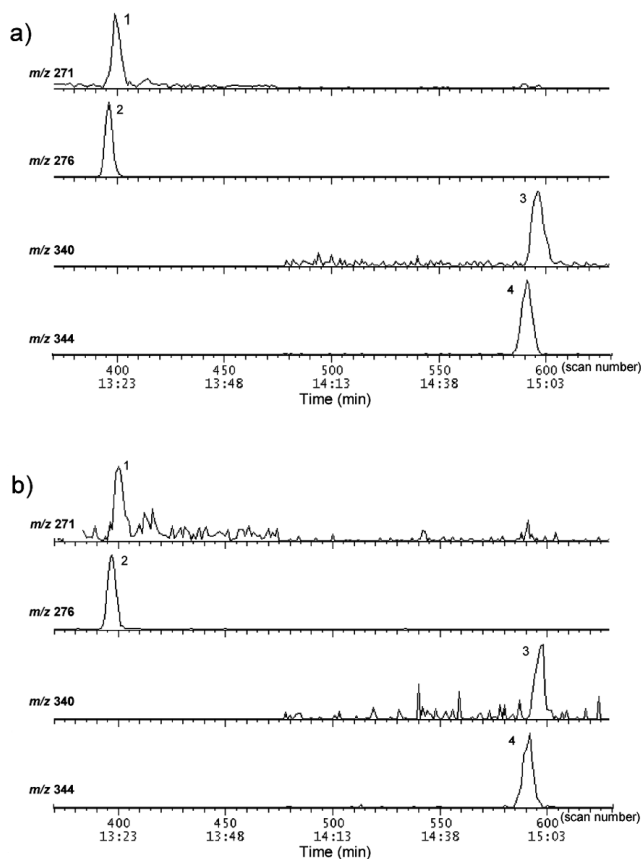
say was lower than 13%, the recovery rate varied in the individual assay (48–68%, n=3). For the assessment of linearity, increasing amounts of E<sub>1</sub> and E<sub>2</sub> (0, 50, 100pg) and ISs (1 ng) were added to the brain homogenate (corresponding to 0.6g of tissue) and assayed, and the typical product ion mass chromatograms were shown in Figure 5. The peak area ratios of the DMES derivatives of E<sub>1</sub> and E<sub>2</sub> toward those of ISs (y) were plotted versus the added amount of E<sub>1</sub> and E<sub>2</sub> (x) to obtain the regression lines (Figures 6a, b). These showed almost satisfactory linearity with correlation coefficient values of more than 0.981 and 0.977, respectively. However, the slope of the lines varied in the individual assay {E<sub>1</sub>:  $7.15 \times 10^{-4}$ , 20.5%; E<sub>2</sub>:  $1.05 \times 10^{-3}$ , 49.7% (mean, RSD, n=4)}, and difference from that of standard sample (E<sub>1</sub>:  $9.00 \times 10^{-4}$ , 4.9%; E<sub>2</sub>:  $8.82 \times 10^{-4}$ , 6.5%, n=3) was observed.

Therefore, the standard addition method was applied for the determination of E<sub>1</sub> and E<sub>2</sub> in rat brains (n=4), and the procedure was shown in Figure 2. The concentration of E<sub>1</sub> and E<sub>2</sub> in rat brains was determined from the distance between the intersection of the line with the x axis and the origin of the coordinates. When the peak corresponding to E<sub>1</sub> or E<sub>2</sub> was not detectable on the chromatogram, without addition of estrogens, the concentration of E<sub>1</sub> or E<sub>2</sub> was expressed as 0 pg/g tissue. The measured brain concentration of E<sub>1</sub> and E<sub>2</sub> were 0, 0, 0, 44.2pg/g tissue and 0, 15.4, 159.0, 37.2 pg/g tissue, respectively.

### 3.3. Determination of guaiacol estrones in rat brains

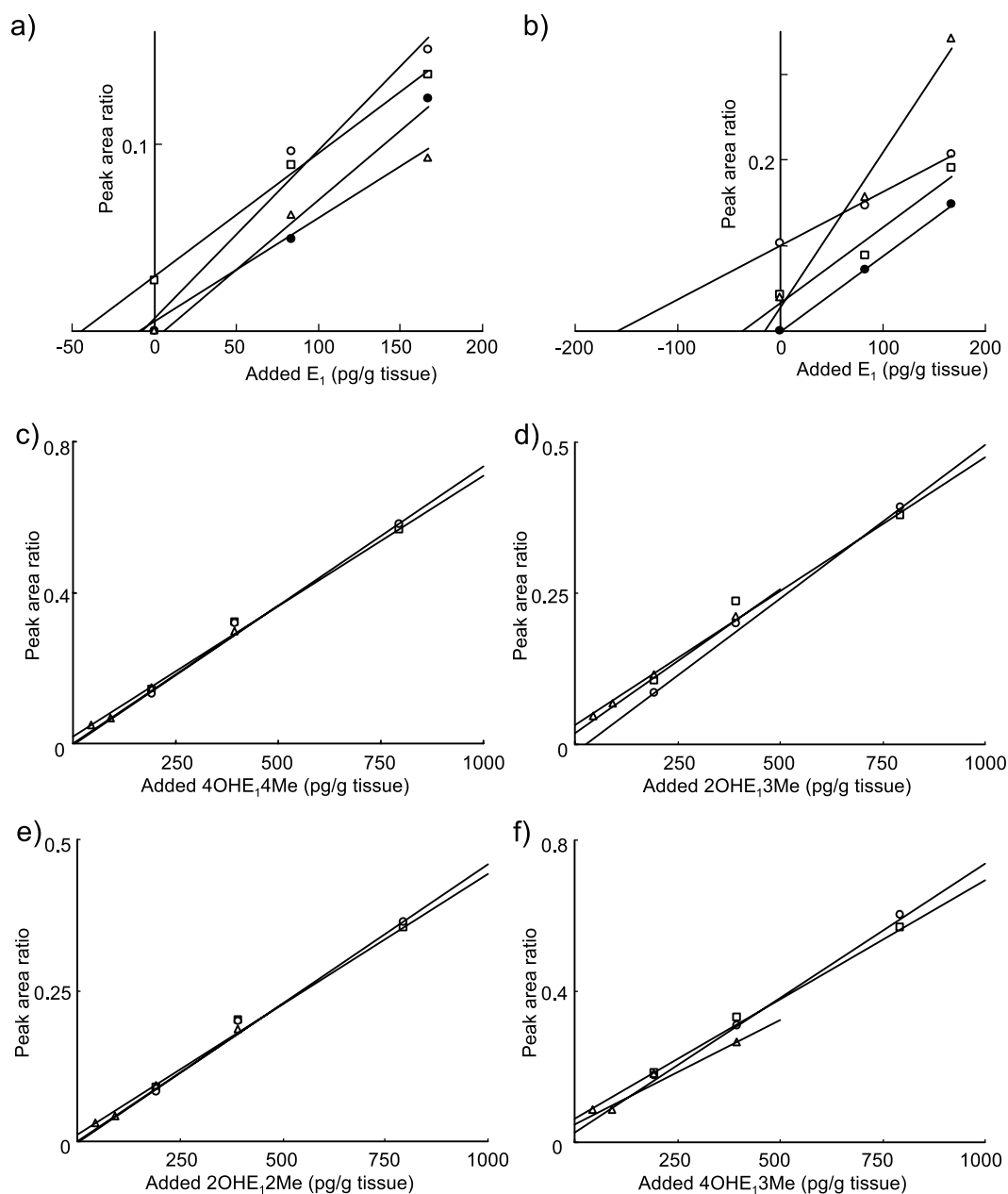
First of all, we attempted to apply the above pretreatment method for determination of guaiacol estrones in order to analyze E<sub>1</sub> and E<sub>2</sub> simultaneously. Authentic samples (0.1–100ng) were subjected to silica-gel mini column chromatography, solid phase extraction, and the elution conditions were examined. Although the silica-gel mini-column chromatography was effective for purification of biological samples, the recovery rates of low amounts of guaiacol estrones (<10ng) were not satisfactory (<10%). These low recovery rates might be caused by irreversible adsorption of the guaiacol estrones on silica-gel, which may depend on the guaiacol structure of the target compound. In the previous paper, we reported that the derivatization to acetate was useful for the treatment of catechol estrogens in the complex matrix and prevention of adsorption [6, 7]. These data prompted us to apply acetylation for determination of guaiacol estrones. Authentic samples (100pg) were acetylated with acetic anhydride-pyridine, subjected to silica-gel mini-column chromatography, solid-phase extraction, and then hydrolyzed with 1% KOH in MeOH. After derivatization with DMESI, the sample was analyzed with GC/MS/MS as shown in Figure 2. The recovery rate was improved and raised higher than 90%.

It was obvious that a stable isotope labeled compound, such as



**Figure 5.** Product ion mass chromatograms of DMES derivatives of E<sub>1</sub>, E<sub>2</sub> (50pg/tube) and ISs (1 ng/tube).

a) Authentic samples, b) extracted E<sub>1</sub>, E<sub>2</sub> and ISs added to rat brain homogenate. Peaks: 1, E<sub>1</sub>; 2,  $[^2\text{H}_5]$ -E<sub>1</sub>; 3, E<sub>2</sub>; 4,  $[^2\text{H}_5]$ -E<sub>2</sub>.



**Figure 6.** Determination of  $E_1$  (a),  $E_2$  (b) and guaiacol estrones (c–f) in rat brains using standard addition method.

**Table 1.** Absolute recovery of guaiacol estrones and IS

Add (pg/tube)	Recovery (%) <sup>a)</sup>				
	2OHE <sub>1</sub> 2Me	2OHE <sub>1</sub> 3Me	4OHE <sub>1</sub> 3Me	4OHE <sub>1</sub> 4Me	16oxoE <sub>2</sub>
0	N.D. <sup>b)</sup>	N.D.	N.D.	N.D.	N.D.
100	65.9	75.5	74.2	65.8	
400	74.3	71.0	70.7	75.7	
(10ng)					63.0

a) Mean (n=3).

b) N.D. = below the detection limit (5 pg/ injection; S/N>5).

the deuteride, was superior for IS in GC/MS analysis; however, the hydrolysis step could cause deuterium–hydrogen exchange. We chose 16oxoE<sub>2</sub> as an IS for analysis of guaiacol estrones, which is not an endogenous compound and is clearly separated from guaiacol estrones in the GC/MS.

To estimate the overall recovery rates, guaiacol estrones (0, 100, and 400pg) and IS (10ng) were added to rat brain homogenate (corresponding to 0.5g of tissue), extracted and treated as above, followed by derivatization and GC/MS/MS analysis. The recovery rates were almost satisfactory taking into consideration the above pretreatment method (Table 1); however, the peaks corresponding to guaiacol estrones were not detectable on the chromatogram of the samples where guaiacol estrones were not added. So the intra-assay reproducibility was examined using brain homogenate to which was added guaiacol estrones (100pg) and IS (10ng). RSD (n=4) of the peak area ratios of 2OHE<sub>1</sub>2Me, 2OHE<sub>1</sub>3Me, 4OHE<sub>1</sub>3Me and 4OHE<sub>1</sub>4Me toward that of IS were 9.4%, 12%, 5.8% and 8.4%, respectively.

In order to assess the linearity, increasing amount of guaiacol estrones (25–400pg) and IS (10ng) were added to the brain homogenate (corresponding to 0.5g of tissue) and assayed. The linearity was satisfactory for all guaiacol estrones; that is, the correlation coefficient values were more than 0.990. Therefore, we attempted to determine guaiacol estrones using the standard addition method, and the obtained regression lines were shown in Figures 6c–f. Although some lines showed the intersection with the y axis, the obtained values were quite smaller than the amount of added standard guaiacol estrones, so that the determination was not reliable. Thus, guaiacol estrones in rat brain were not determined by using the method.

#### 4. Discussion

We developed a method to determine E<sub>1</sub>, E<sub>2</sub> and guaiacol estrones in rat brains using GC/MS/MS and the standard addition method. The standard addition method needs large amount of sample and is somewhat tedious; however, the method is required to remove the effect of endogenous substances especially for analyzing the low amount of compounds in a complex matrix such as the brain.

Rahimy *et al.* reported the E<sub>2</sub> value in rat brain is *ca.* 150–290pg/g tissue [10], which is measured with radioimmunoassay and compatible with our results. But E<sub>1</sub> and E<sub>2</sub> levels in rat brains were not much higher than those in rat serum (*ca.* 140–240pg/ml, *ca.* 100–150pg/ml, respectively) [11], so it is ambiguous that estrogens act as neurosteroids [1]. The individual differences were also observed in male rat brains regarding E<sub>1</sub> and E<sub>2</sub>, although estrogen levels in serum of male rats do not fluctuate unlike those of

female rats which have an estrous cycle. Additional experiments are required in order to clarify this ambiguity, and the developed method would be helpful to elucidate the relationship between estrogen levels and the neural state.

Although we reported the existence of guaiacol estrones using two or three whole brains [5], the developed method using a small sample of the brain could not give the reliable determination data of guaiacol estrones. However, we have already demonstrated the existence of hydroxylase and catechol–O–methyltransferase in the brain to give catechol estrogens and guaiacol estrogens, respectively [12]. The development of a more sensitive determination method, such as immunoassay, is necessary to overcome this discrepancy.

#### Acknowledgement

We thank Teikoku Hormone Mfg. and Nippon Kayaku for supplying estrogens. This work was supported in part by a grant from the Ministry of Education, Science, Sports and Culture of Japan.

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